

Nanocrystalline materials for high temperature soft magnetic applications: A current prospectus

M E McHENRY*, M A WILLARD, H IWANABE[†], R A SUTTON[‡], Z TURGUT, A HSIAO and D E LAUGHLIN

Materials Science and Engineering, Carnegie Mellon University, Pittsburgh, PA 15213-3890, USA

[†]Currently at Japan Electric Corporation, Japan

[‡]Currently at University of Queensland, Australia

Abstract. Conventional physical metallurgy approaches to improve soft ferromagnetic properties involve tailoring chemistry and optimizing microstructure. Alloy design involves consideration of induction and Curie temperatures. Significant in the tailoring of microstructure is the recognition that the coercivity, (H_c) is roughly inversely proportional to the grain size (D_g) for grain sizes exceeding $\sim 0.1-1 \mu\text{m}$ (where the grain size exceeds the Bloch wall thickness, δ). In such cases grain boundaries act as impediments to domain wall motion, and thus fine-grained materials are usually harder than large-grained materials. Significant recent development in the understanding of magnetic coercivity mechanisms have led to the realization that for very small grain sizes $D_g < \sim 100 \text{ nm}$, H_c decreases sharply with decreasing grain size. This can be rationalized by the extension of random anisotropy models that were first suggested to explain the magnetic softness of transition-metal-based amorphous alloys. This important concept suggests that nanocrystalline and amorphous alloys have significant potential as soft magnetic materials. In this paper we have discussed routes to produce interesting nanocrystalline magnets. These include plasma (arc) production followed by compaction and primary crystallization of metallic glasses. A new class of nanocrystalline magnetic materials, HITPERM, having high permeabilities at high temperatures have also been discussed.

Keywords. Soft magnetic materials; nanocrystals; NANOPERM, HITPERM, high temperature magnetic materials.

1. Introduction

Over several decades amorphous and, more recently, nanocrystalline materials have been investigated for applications in magnetic devices requiring magnetically soft materials, such as transformers, inductive devices, etc. Chemical and structural variations on a nanoscale are important for determining optimal magnetic properties. In this paper we shall discuss recent developments in the synthesis, structural characterization, properties and applications of nanocrystalline (and amorphous) magnets. We have considered processing routes to control chemistry and microstructural morphology on increasingly smaller length scales, and developing experimental techniques which allow more accurate and quantitative probes of structure (including magnetic domain structure) on smaller length scales.

Demands on bulk soft magnetic materials include higher combined induction and permeabilities, and for some applications magnets capable of operating at higher tem-

peratures as well as many other nonmagnetic issues such as mechanical properties, corrosion resistance, etc. For achieving such goals, key issues include alloy chemistry, structure, and more importantly the ability to tailor microstructural features. It is clear that the magnets used in soft magnetic applications must be optimized in terms of their intrinsic and extrinsic magnetic properties as well as their morphology. The key intrinsic magnetic properties, the saturation magnetic induction, B_s , and Curie temperature, T_c , however, are determined solely by alloy composition.

Premiere high induction bulk materials include Fe, Fe-Si, Co and FeCo alloys (for reviews see Chen 1986; Boll 1994). FeCo bulk alloys (Pfeifer and Radloff 1980; Rajkovic and Buckley 1981) are particularly attractive because of their high inductions, while their permeabilities are smaller than those of amorphous and nanocrystalline alloys and Si steels. The extrinsic property of interest is the magnetic permeability, μ , the magnetic response function in an applied field which is usually inversely related to the materials coercivity, H_c . The magnetic permeability, μ , is determined by chemistry, structure and morphology (shape). In particular, alloys with small

*Author for correspondence

magnetocrystalline anisotropies and magnetostrictive coefficients give rise to particularly soft magnetic materials.

Choice of soft magnetic materials for applications has been guided by recent developments in the field of soft magnetic materials as summarized in figure 1. Amorphous and nanocrystalline magnetic materials, in terms of combined induction and permeabilities are now competitive with Si-Fe bulk alloys, and the above-mentioned Fe-Co alloys. In figure 1a (adapted from Makino *et al* 1997),

figures of merit, B_s and μ_e , for Co-based amorphous alloys, Fe-based amorphous alloys, and nanocrystalline alloys are summarized, which have evolved over the past decades with soft magnetic properties exceeding those of the bulk alloys based on Fe, Co, and Fe-Co. Nanocrystalline Fe-Si-B-Nb-Cu alloys have been patented under the trade name FINEMETS[®]TM (Yoshizawa *et al* 1988a, b; Yoshizawa and Yamauchi 1989, 1990, 1991). Soft materials, based on FeMBCu, have been patented under the trade name NANOPERM[®]TM. These Fe-M-B (M=Zr, Nb, Hf, ...) nanocrystalline alloys have all been optimized to achieve small magnetostrictive coefficients with concomitant large permeabilities. More recently (Fe, Co)MBCu (M=Nb, Hf, or Zr) nanocrystalline alloys, called HITPERM have been shown to have attractive inductions (1.6–2.1 T) combined with high permeabilities and high Curie temperatures. In FINEMETS α -FeSi nanoparticles with a DO₃ structure are observed, in NANOPERM α -Fe particles with a bcc structure are formed. In HITPERM alloys (Willard 1998, 1999, to appear) nanocrystalline α -FeCo are formed with significantly improved high temperature magnetic properties than in the former two.

Conventional physical metallurgy approaches to improve soft ferromagnetic properties involve tailoring chemistry and optimizing microstructure. Significant in the tailoring of microstructure is the recognition that a measure of the magnetic hardness (the coercivity, H_c) is roughly inversely proportional to the grain size (D_g) for grain sizes exceeding ~ 0.1 – $1 \mu\text{m}$ (where the grain size exceeds the Bloch wall thickness, δ). In such cases grain boundaries act as impediments to domain wall motion, and thus fine-grained materials are usually harder than large-grained materials.

Significant recent development in the understanding of magnetic coercivity mechanisms have led to the realization that for very small grain sizes $D_g < \sim 100 \text{ nm}$, (Herzer and Hilzinger 1986, 1989; Herzer 1990, 1992, 1995, 1997), H_c decreases quickly with decreasing grain size (figure 1b). This can be rationalized by the extension of random anisotropy models (Alben *et al* 1978a, b) that were first suggested to explain the magnetic softness of transition-metal-based amorphous alloys. This is illustrated by the fact that the domain wall, whose thickness δ , which exceeds the grain size, now samples several or many grains so that fluctuations in magnetic anisotropy on the crystal length scale are irrelevant to domain wall pinning. This important concept suggests that nanocrystalline and amorphous alloys have significant potential as soft magnetic materials. Soft magnetic properties require that nanocrystalline grains be exchange coupled, and therefore any of the processing routes yielding free-standing nanoparticles must be coupled with a compaction method in which the magnetic nanoparticles end up exchange coupled.

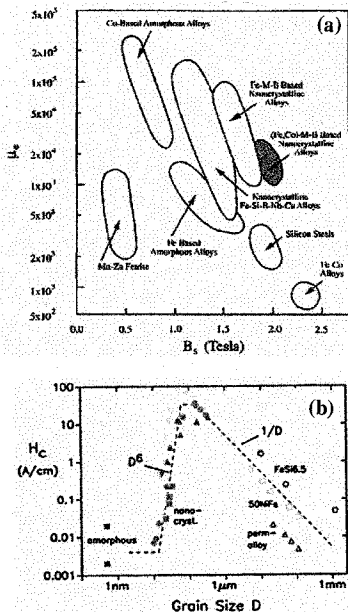


Figure 1. (a) (Adapted from Makino *et al* 1997). Relationship between permeability μ_e (at 1 kHz) and saturation polarization for soft magnetic materials and (b) Herzer diagram illustrating dependence of the coercivity, H_c , with grain size in magnetic alloys.

In this paper we shall explore and illustrate issues that are pertinent to the general understanding of the magnetic properties of nanocrystalline materials. The development of nanocrystalline materials for soft magnetic applications is an emerging field for which we will try to offer a current perspective that is expected to evolve further with time.

2. Alloy design

Alloy design includes issues of chemistry and processing designed to optimize one of a number of important intrinsic and/or extrinsic magnetic properties as well/or to optimize structural or microstructural features which promote important (usually extrinsic) magnetic properties. The first of these issues concerns the choice of chemistry so as to have an impact on the intrinsic magnetization of the material. The second issue involves alloy additions designed to aid in the formation of an amorphous or bulk amorphous phase, as a means to an end or as a precursor to producing a nanocrystalline material.

2.1 Intrinsic magnetic properties

The desire for large inductions limits choices of alloys to those rich in Fe or Co and therefore near the top of the Slater–Pauling curve. As described in the Slater–Pauling curve, Fe–Co alloys exhibit the largest magnetic inductions on any material, and also have T_c 's desirable for high temperature applications. Alloys near the equiatomic composition are particularly soft and exhibit large permeabilities, but this magnetic softness is rooted in a zero crossing of the first-order magnetic anisotropy constant, K_1 . Fe-rich alloys typically have smaller inductions and lower Curie temperatures than Fe–Co alloys. For applications such as rotor assemblies in the more electric aircraft (Quigley) where operation at temperatures up to 600°C are desired no other alloy system offers significant potential for these applications. For this reason our group has concentrated on nanocrystalline alloys based on FeCo.

R A Sutton (unpublished) has used LKKR band structure calculations to determine local moments on Fe and Co in the disordered and ordered B2 $\text{Fe}_{1-x}\text{Co}_x$ alloys. The local moment on Fe was found to be sensitive to composition, while that of Co is relatively invariant with composition (figure 2a). The net alloy moment reproduces the trends of the famous Slater–Pauling curve. An approximately 4% larger alloy moment is observed for equiatomic FeCo in the ordered B2 (CsCl) structure. Mean field theory fits to temperature-dependent magnetization data for equiatomic FeCo (figure 2b), predicts that equiatomic FeCo would have a Curie temperature of 1475 K if it could be stabilized with respect to the $\alpha \rightarrow \gamma$ phase transition. This large T_c for FeCo leads to a very flat $M(T)$ at temperatures of interest for the MORE electric plane (550–600°C).

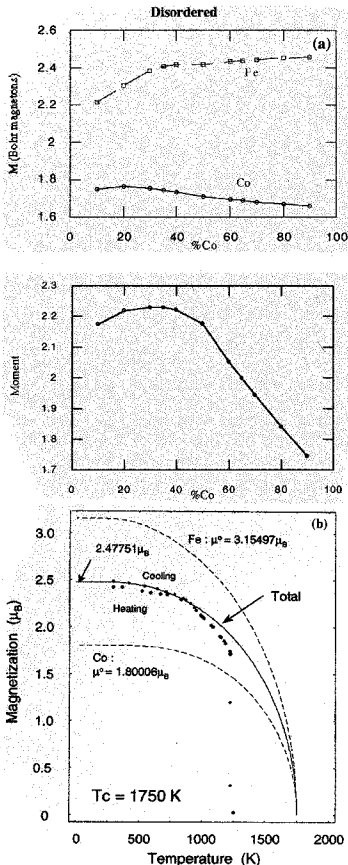


Figure 2. (a) Spin-only Slater–Pauling curve for a disordered Fe–Co alloy as determined from LKKR band structure calculations (R A Sutton, unpublished data), Co-site, Fe-site and total moment as a function of composition and (b) mean field theory fit to temperature dependent magnetization data for equiatomic FeCo using moments of (a).

2.2 Considerations of glass forming ability and crystallization

The considerations for synthesizing amorphous metallic alloys are often at odds with the considerations for optimizing magnetic properties. This is because the addition of typical glass forming agents has a deleterious effect on such intrinsic magnetic properties as the saturation induction and the Curie temperature. Considerations for glass forming ability in binary systems include the temperature and composition of the eutectic, the terminal phases to the right and left of the eutectic composition, the slope of the liquidus as a function of composition, etc. If the amorphous alloy is to be used as a precursor for the production of a nanocrystalline material, the primary and secondary crystallization temperatures are of importance as is the structure of the ferromagnetic nanocrystalline phase that is the product of primary crystallization. Table 1 summarizes these parameters for selection of binary Fe-early transition metal (TE) and metalloid (M) systems. Significant alloy systems include Fe(Co)-TE-B-Cu alloys wherein early transition metals (TE) and B are added as glass formers, and where Cu is added to provide nanocrystal nucleation sites. The low TE eutectic concentration in Fe(Co)-TE systems is fortuitous in that it implies relatively small moment reduction in these alloys. In crystallization of these alloys, α -Fe or α' -FeCo nanocrystals are thought to nucleate on Cu clusters expelling Zr and B to an intergranular amorphous phase. Such systems form the basis for FINEMET, NANOPERM, and HITPERM alloys.

3. Synthesis

Common synthesis routes for producing amorphous and/or nanocrystalline materials for soft magnetic applications which are viable for producing bulk alloys for larger-scale applications, as opposed to thin film synthesis for microelectronic applications, include the following routes (though others exist) for example. These synthesis routes are:

(I) *Powder synthesis techniques*: Magnetic nanocrystals may be synthesized as free-standing powders or nanocapsulates, which must then be compacted to form a

bulk alloy with nanocrystalline grains. Selected powder synthesis techniques that include C-arc synthesis (McHenry *et al* 1994, 1996, 1998), plasma torch synthesis, and mechanical milling techniques.

(II) *Rapid solidification processing*: Amorphous alloys can be produced by a variety of rapid solidification processing routes. These typically require cooling rates of $> 10^4$ K/s for typical alloys at a eutectic composition. Slower cooling rates are possible for bulk amorphous alloys.

(III) *Solidification processing of bulk amorphous alloys*: Bulk amorphous alloys are formed by more conventional solidification routes with slower cooling rates. The so-called large glass forming abilities of these alloys allow for production of amorphous materials with much larger dimensions, thus the name bulk amorphous alloys.

(IV) *Crystallization of amorphous precursors*: Nanocrystalline bulk alloys with nanocrystalline grains can also be produced by solid state reaction (crystallization) of an amorphous precursor. Our group at CMU has concentrated on routes (I), (II) and (IV) for the synthesis of nanocrystalline magnets. Plasma torch synthesis produces free-standing nanoparticles which must be followed by compaction to produce bulk magnets. Melt spinning produces an amorphous precursor which is then nanocrystallized to yield a two-phase nanocrystalline/amorphous magnetic material.

4. Structure and magnetic properties

Gallagher *et al* (1996) have produced $\text{Fe}_2\text{Co}_{1-x}\text{[C]}$ ($x=0.0, 0.2, 0.4, 0.5, 0.6, 0.8$, nominally) free-standing nanoparticles by a Kratschmer-Huffman carbon arc method. X-ray diffraction indicated a single-disordered bcc FeCo phase (along with graphitic C), for all but the $x=0.2$ composition for which some fcc Co was observed. Particles had an average diameter of 50 nm from Scherrer analysis of the peak widths, and a broad size distribution based on TEM. Imaging energy-filtered microscopy showed Fe and Co to be uniformly distributed in the nanoparticles. A compositional dependence of the $M(H=1\text{ T})$ (figure 3b) magnetization was similar to that observed in the bulk alloy Slater-Pauling curve. $\text{Fe}_{0.5}\text{Co}_{0.5}\text{[C]}$ exhibited the largest magnetization heretofore

Table 1. Considerations for select glass forming ability in binary systems: temperature and composition of the eutectic, the slope of the liquidus, and the terminal phases to the right and left of the eutectic composition.

Binary alloy	x_e at% (wt%)	T_E (C)	dT_L/dx	Terminal phases	
Fe-B	17 (3-8)	1174	21-4	Fe	Fe ₂ B
Fe-Si	33 (20)	1212	~ 10	Fe	Fe ₂ Si(Fe ₃ Si)
Fe-Zr	10 (15)	1337	20	Fe	Fe ₂ Zr(Fe ₃ Zr)
Fe-Hf	8.5 (22)	1390	18	Co	λ
Fe-Nb	12 (18.5)	1373	~ 14	Fe	ϵ
Fe-Ta	7.9 (21)	1442	12-5	Fe	ϵ

observed in similarly produced nanoparticles. The $M(H)$ curves did not saturate, presumably due to some fraction of superparamagnetic particles, or from canted interfacial spins. These FeCo containing soots, reported by Gallagher *et al* (1996), had over 200 emu/g magnetizations (note that α -Fe has a ~ 220 emu/g saturation magnetization) (figure 3a).

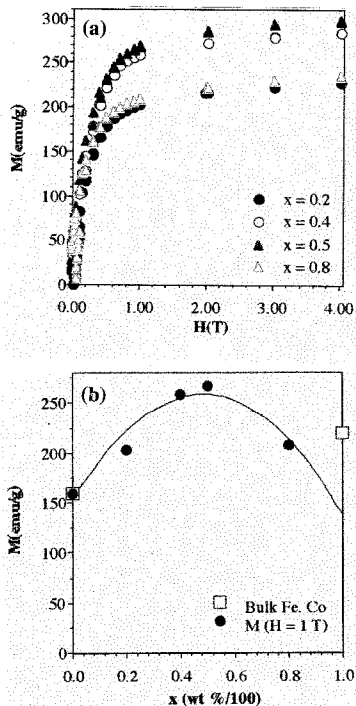


Figure 3. (a) 5 K $M(H)$ of $\text{Fe}_x\text{Co}_{1-x}[\text{C}]$ nanoparticles ($x = 0.2, 0.4, 0.5, 0.6$, wt%/100) and (b) $M(H = 1 \text{ T})$ vs x for nanoparticles including data for bulk Fe and Co for comparison. Parabola is a guide to the eye (reproduced from Gallagher *et al* 1996).

Fe-Co alloys undergo an order-disorder phase transformation at a maximum temperature of 725°C at the composition $\text{Fe}_{50}\text{Co}_{50}$ with a change in structure from the disordered α -bcc (A1) to the ordered α' -CsCl(B2)-type structure (figure 4a). C-coated $\text{Fe}_x\text{Co}_{1-x}$ ($x = 0.50, 0.45, 0.40, 0.35, 0.30, 0.25$) nanoparticles have been produced using a RF plasma torch by Turgut *et al* (1997, 1998). The only C source was acetylene used as a carrier gas. Structural determination by X-ray diffraction indicated a single disordered bcc α -FeCo phase along with graphitic C for all compositions. EDX analysis (Scott *et al* 1997) indicated compositional fluctuations of a few atomic % for individual particles of one nominal composition, which is attributed to starting with elemental rather than alloy precursors. Magnetic hysteresis loops have been measured to $T > 1050$ K and revealed relatively high room temperature coercivities (200–400 Oe), with a strong compositional variation similar to that observed in bulk alloys. Larger coercivities are consistent with particles near the monodomain size for these alloys. The T -dependence of the magnetization revealed the effects on atomic ordering. Figure 4b illustrates $M(T)$, at $H = 500$ Oe, for equiatomic FeCo nanocrystalline powders measured on a second heating cycle after initial heating to 920°C. The influence of ordering is observed prominently in features that are quite similar to thermomagnetic observations for bulk $\text{Fe}_{49}\text{Co}_{50}\text{V}_2$ alloys. These features include a discontinuity in $M(T)$ at $\sim 600^\circ\text{C}$ (due to chemical ordering) and a return to the extrapolated low temperature branch of the curve at the disordering temperature of $\sim 730^\circ\text{C}$.

In general, the mean field theory describing Curie temperatures for nanoparticles is not strongly deviate from that of the bulk materials. However, this may not be true for very small nanoparticles (< 5 nm) where the fraction of surface atomic sites is large, and therefore the reduced coordination of surface atoms should in fact alter the Curie temperature. For most observations in the literature, however, the T_c does not deviate strongly from that of the bulk materials. Of note in figure 4b is that at $\sim 950^\circ\text{C}$ we observe an abrupt drop in the magnetization that corresponds to the $\alpha \rightarrow \gamma$ structural phase transformation. It can be concluded from the abruptness in the drop of $M(T)$ that the $\text{Fe}_{0.5}\text{Co}_{0.5}$ alloy has a Curie temperature that exceeds the $\alpha \rightarrow \gamma$ phase transformation temperature. This is corroborated by the differential thermal analysis data, where the order-disorder and $\alpha \rightarrow \gamma$ phase transformations have also been clearly observed.

NANOPERM and HITPERM alloys have been synthesized by Willard *et al* (1998, 1999 to appear) through nanocrystallization of amorphous precursors. Figure 5a shows observations of magnetization as a function of temperature (Willard *et al* 1998) for two alloys, one of a NANOPERM composition and the other of a HITPERM

composition. Initially, the two samples are amorphous, and as the samples are heated the crystallization of the amorphous phase becomes apparent in each alloy. The NANOPERM material has its Curie temperature just above room temperature. The magnetic phase transition is followed by primary crystallization at $T_{c1} \sim 500^\circ\text{C}$, which in turn is followed by secondary crystallization, and then finally the magnetic phase transition for the α -Fe nanocrystalline phase at $\sim 770^\circ\text{C}$. Thus re-entrant ferromagnetism is observed in amorphous alloys with primary crystallization temperatures above the Curie temperature of the amorphous phase.

Crystallization kinetics of NANOPERM have been studied by Hsiao *et al* (1998, unpublished) who examined

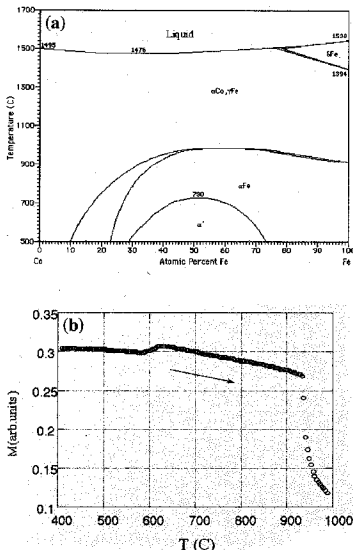


Figure 4. (a) Fe-Co phase diagram (produced using TAPP@TM software, ES Microwave) and (b) $M(T)$ data for C-arc synthesized equiatomic Fe-Co nanoparticles measured using a Lakeshore model 7300 VSM and oven assembly at 500 Oe measured on a second heating cycle to 995°C after initial heating to 920°C (Turgut *et al* 1997).

isothermal time-dependent magnetization at temperatures above the crystallization temperature (figure 5b). Since the amorphous phase is paramagnetic at the crystallization temperature, the magnetization is a direct measure of the volume fraction of the α -Fe crystalline phase that has transformed. $M(T)$ then measures the crystallization kinetics. Figure 5b shows curves reminiscent of Johnson-Mehl-Avrami kinetics for phase transformation.

It is noted that in nanocrystalline materials, with two phase microstructures, the Curie temperature of the nanocrystalline phase and that of the intergranular phase are both important parameters in describing the magnetic response. The intergranular phase, which is typically richer in nonmagnetic species, can have a suppressed

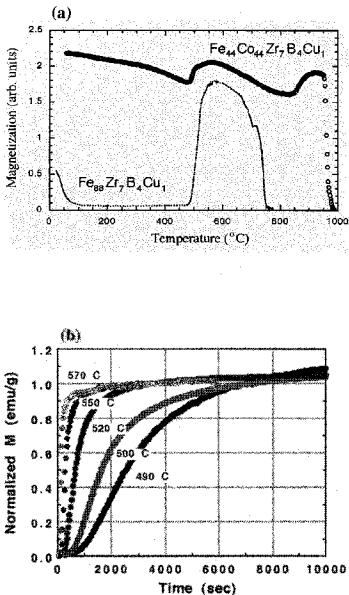


Figure 5. (a) $M(T)$ (Willard *et al* 1998) for an alloy with a NANOPERM composition $\text{Fe}_{44}\text{Co}_{44}\text{Zr}_7\text{B}_4\text{Cu}_1$ and an alloy with a HITPERM composition $\text{Fe}_{88}\text{Co}_{44}\text{Zr}_7\text{B}_4\text{Cu}_1$, and (b) isothermal magnetization as a function of time for the NANOPERM composition $\text{Fe}_{88}\text{Zr}_7\text{B}_4\text{Cu}_1$ at 490, 500, 520, 550 and 570°C , respectively (Hsiao, unpublished data).

magnetic order parameter and lower Curie temperature than the nanocrystals. This is especially important in describing the temperature dependence of a variety of extrinsic magnetic properties. In figure 5a the $M(T)$ curve for HITPERM, shows that the magnetization decreases monotonically until $\sim 400^\circ\text{C}$ as the amorphous phase approaches its Curie temperature. Above $400\text{--}500^\circ\text{C}$ structural relaxation and crystallization of the α' -FeCo phase occurs resulting in larger magnetization due to the larger Curie temperature of the α' -FeCo phase. The crystallization temperature is apparently well below the Curie temperature of the amorphous phase, so that the magnetization of the amorphous phase is only partially suppressed prior to crystallization. At a temperature corresponding to the $\alpha\text{--}\gamma$ phase transition (980°C), the material abruptly loses its magnetization, consistent with the paramagnetic response of the γ phase.

Permeability measurements were made on a HITPERM sample that was prepared as a laminate in a toroidal geometry, and annealed at 600°C . The frequency dependence of the real and imaginary components of the complex permeability, μ' and μ'' , reflect the power loss due to eddy currents and hysteretic response. The maximum permeability for this material was determined to be about 1800. $\mu''(T)$ peaks at a frequency of ~ 20 kHz. The high peak frequency is thought to reflect the higher resistivity in the nanocrystalline materials compared with conventional bulk alloys. The resistivity is a significant term in eddy current-related damping of domain wall motion.

5. Conclusions

FeCo alloys possess large inductions and Curie temperatures, suggesting their use in high-temperature soft-magnetic applications. Free-standing FeCo nanoparticles with large inductions and flat $M(T)$ to $\sim 970^\circ\text{C}$ have been produced by C-arc and plasma torch routes. A new nanocrystalline soft magnetic material, produced by nanocrystallization of an amorphous precursor, with the α' -FeCo phase (B2 structure) are described with excellent intrinsic magnetic properties and strong intergranular coupling to 600°C . These have excellent soft magnetic properties to high frequencies and temperatures due to the nanocrystalline grains. This material is therefore a candidate for high temperature aircraft power applications.

Acknowledgements

MEM thanks the National Science Foundation for support through Grant No. ECD-8907068. Efforts have also been sponsored by the Air Force Office of Scientific Research, Air Force Materiel Command, USAF, under grant number F49620-96-1-0454. The authors acknowledge useful discussions with Prof. S A Majetich, Dr J H Scott, Dr V

Harris, Dr B Lu, Ms K Gallagher, Dr S-Y Chu, and Dr M-Q Huang.

References

- Alben R, Budnick J and Cargill III G S 1978a in *Metallic glasses* (Ohio: ASM) p. 304
- Alben R, Becker J J and Chi M 1978b *J. Appl. Phys.* **49** 1653
- Boll R 1994 *Soft magnetic metals and alloys, in Materials science and technology: A comprehensive treatment* (ed. K H J Buschow (Weinheim: VCH) Ch. 14, Vol. **3B**, p. 399
- Chen C W 1986 *Magnetism and metallurgy of soft magnetic materials* (New York: Dover Publications Inc.)
- Gallagher K, Johnson F, Kirkpatrick E, Scott J H, Majetich S and McHenry M E 1996 *IEEE Trans. Mag.* **32** 4842
- Herzer G 1990 *IEEE Trans. Magn.* **26** 1397
- Herzer G 1991 *Mater. Sci. & Eng.* **A133** 1
- Herzer G 1992 *J. Magn. Magn. Mater.* **112** 258
- Herzer G 1995 *IEEE Trans. Mag.* **26** 1397
- Herzer G 1997 *Handbook of magnetic materials* (ed.) K H J Buschow (Amsterdam: Elsevier Science) Ch. 3, Vol. **10**, p. 415
- Herzer G and Hiltzinger H R 1986 *J. Magn. Magn. Mater.* **62** 143
- Herzer G and Hiltzinger H R 1989 *Phys. Scr.* **39** 639
- Makino A, Hatanai T, Naitoh Y, Bitoh T, Inoue A and Masumoto T 1997 *IEEE Trans. Magn.* **33** 3793
- McHenry M E and Subramoney S 1998 *Fullerenes: Physics and technology* (eds) K M Kadish and R S Ruoff (New York: John Wiley) (to appear in 1999)
- McHenry M E, Majetich S A, De Graef M, Artman J O and Staley S W 1994 *Phys. Rev.* **B49** 11358
- McHenry M E, Gallagher K, Johnson F, Scott J H and Majetich S A 1996 in *Recent advances in the chemistry and physics of fullerenes and related materials, ECS symp. proceedings, PV 96-10, Pennington, NJ* (eds) K M Kadish and R S Ruoff (N.J.: Pennington) p. 703
- Pfeifer F and Radloff C 1980 *J. Magn. Magn. Mater.* **19** 190
- Rajkovic M and Buckley R A 1981 *Metal Sci.* 21
- Scott J H, Majetich S A, Turgut Z, McHenry M E and Boulos M 1997 *Carbon coated nanoparticle composites synthesized in an RF plasma torch, in Nanostructured materials, MRS proc.* (ed.) J C Parker (Pittsburgh, PA: Materials Research Society) (in press)
- Turgut Z, Huang M-Q, Gallagher K, Majetich S A and McHenry M E 1997 *J. Appl. Phys.* **81** 4039
- Turgut Z, Scott J H, Huang M-Q, Majetich S A and McHenry M E 1998 *J. Appl. Phys.* **83** 6468
- Willard M A, Laughlin D E, McHenry M E, Thoma D and Sickafus K 1998 *J. Appl. Phys.* (to appear)
- Willard M A, Laughlin D E and McHenry M E 1999 *J. Appl. Phys.* (to appear)
- Yoshizawa Y and Yamauchi K 1989 *J. Jap. Inst. Metals* **53** 241
- Yoshizawa Y and Yamauchi K 1990 *Mater. Trans. JIM* **31** 307
- Yoshizawa Y and Yamauchi K 1991 *Mater. Sci. & Eng.* **A133** 176
- Yoshizawa Y, Oguma S and Yamauchi K 1988a *J. Appl. Phys.* **64** 6044
- Yoshizawa Y, Yamauchi K, Yamane T and Sugihara H 1988b *J. Appl. Phys.* **64** 6047